

Title	Ore mineralogy and mineral chemistry of pyrite, galena, and sphalerite at Soripesa Prospect area, Sumbawa island, Indonesia
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ORE MINERALOGY AND MINERAL CHEMISTRY OF PYRITE, GALENA, AND SPHALERITE AT SORIPESA PROSPECT AREA, SUMBAWA ISLAND, INDONESIA

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Abstract

The Soripesa prospect area is located at Maria village, Wawo district, Bima region in the East Sumbawa Island, Indonesia. Lithology is dominantly composed of a lithic-crystal tuff of andesitic and dacitic composition and bedded limestone. The polymetallic epithermal quartz veins are hosted by andesitic volcanoclastic rocks. Within these veins, multiphases, colloform-crustiform, bedding to massive textures with pyrite, sphalerite, galena, chalcopyrite, chalcocite, azurite, and malachite are observed. Selected samples were analyzed by using ore microscopy and SEM-EDX. Ore minerals show replacement, ex-solution, colloform, and zonal textures. The paragenesis diagram was made from a careful study of polished sections and thin sections. Textures of ore minerals such as banded, exsolution, replacement, and zone, have been interpreted to correspond to the order of deposition. In pyrite, the average content of Co (0.45 wt.%) is higher than Ni content (0.14 wt.%) and it means that their origin may be hydrothermal origin. Average content ratio, Co:Ni is 2.81. Galena shows a low Ag content of 0.07 % in average. But they show a high Au content of 1.48 % in average. Sphalerite shows a low Fe content of 1.04 % in average and occasionally chalcopy-

rite inclusion/disease also occurred. Ga and Ge contents are also high in sphalerite. $Co > Ni$ in pyrite, low content of Ag in Galena, low content of Fe and mole % FeS in sphalerite, high content of Ga and Ge, and $\log(Ga/Ge)$ in sphalerite, show that pyrite, galena, and sphalerite from Soripesa prospect area were formed under low temperature condition of hydrothermal fluid.

Keywords: Ore textures, paragenesis, deposition, hydrothermal fluid, low temperature

1 Introduction

The Soripesa prospect area is located in Maria village, Wawo district, Bima regency, West Nusa Tenggara Province. The prospect area is previously owned by PT Bima Baruna Raya Mining (BBRM) and PT Sumbawa Timur Mining. These companies have observed Au-Ag deposit and base metal mineralization. At present, PT Bima Putera Minerals (Indomining Group) has a Mining Permit (exploration stage) for those deposits. The prospect area is mainly composed of andesitic and dacitic volcanoclastic host rocks and small amount of Tertiary bedded limestones. There have five main veins in the Soripesa prospect area, trending nearly north-south. Those veins are hosted in andesitic volcanoclastic rocks (lower Miocene in age). The common ore minerals

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of those veins are chalcopyrite, galena, sphalerite, pyrite, chalcocite, covellite, azurite, and malachite, which are forming as polymetallic epithermal quartz veins.

The purpose of this paper is to identify the ore mineral paragenesis and ore forming processes, and mineral chemistry of some sulfide minerals, especially for pyrite, galena, and sphalerite from polymetallic epithermal quartz veins at Soripesa prospect area, East Sumbawa, Indonesia.

2 Regional Geology

The Sumbawa Island forms as a part of the Cenozoic calc-alkaline volcanic inner Banda-Sunda Arc which is still active up to present. Sunda-Banda island arc is a volcanic arc formed by the interaction of plate subduction slab in the form of Indo-Australia with Asian plate (Carlile and Mitchell, 1994). Sunda-Banda arc is the longest in Indonesia, extending from north Sumatra through Java, Bali, Lombok, and Sumbawa, to east Damar. The shape of the island arc is now being modified in the east due to collision with the Australian–New Guinea continental margin, including East Sumbawa (Hamilton, 1974).

The Sumbawa Island is mainly composed by andesitic to basaltic lava and breccia of the Lower Miocene, with intercalations of tuff and limestone, fresh pyroclastic sequences and shoshonite. This sequence is overlain in parts by dacitic tuff and bedded limestone of the Mid Miocene. These units have been intruded by numerous small to medium bodies (several km's) in the Middle to Upper Miocene including andesite, dacite, diorite, trachyte and syenite (Figures 1 and 2). Sumbawa Island, regionally, is intersected by NW-SE and NE-SW trending structures. However, the formation of quartz veining, alteration and mineralization at Soripesa Prospect are related to the N-S faulting (Noya *et al*, 2009).

3 Sampling and Analytical Methods

Eighteen selected samples were made thin sections and polished sections to identify the ore

minerals and their textures. Ore textures and ore mineral assemblages are analysed under ore microscopy at Earth Resource Engineering Department, Kyushu University. An Energy Dispersive X-Ray Analyser (EDX or EDA) is also used to provide elemental identification and quantitative compositional information. For SEM-EDX analyses, 15 polished sections and 3 thin sections were used to identify the mineral chemistry by using a SHIMADZU SS-550 SEM with a Genesis 2000 energy dispersive spectrometer (EDX) at the Center of Advanced Instrumental Analysis, Kyushu University. Semi-quantitative analyses were performed at an accelerating potential of 15 kV, with a beam current of about 6.0×10^{-10} A, and a beam diameter of $3 \mu\text{m}$. All the data were corrected with a ZAF matrix-correction program. The lowest detection limit is 100 ppm. Ore minerals such as sphalerite, galena, pyrite, chalcopyrite, chalcocite, azurite, malachite, oxide minerals, and some unknown minerals were analyzed by SEM-EDX for elemental identification and compositional information. Based on the elemental compositions, ore minerals can be estimated and identified their ore genesis, and ore forming processes.

4 Results and Discussions

4.1 Ore Minerals Assemblages and Their Textural Characteristics

Common ore minerals of the polymetallic epithermal quartz veins at Soripesa prospect area are chalcopyrite [CuFeS_2], galena [PbS], sphalerite [ZnS], azurite, malachite, chalcocite, covellite, pyrite [FeS_2], hematite, and other iron-oxide minerals.

Chalcopyrite. chalcopyrite is one of abundant ore minerals and also occurs within sphalerite as chalcopyrite disease/inclusion (Figure 3e). This texture is called as ex-solution texture. Some chalcopyrite replace in pyrite (Figure 3g). The cracking of some chalcopyrite are also replaced by Fe oxide minerals such as colloform goethite and hematite. Silver inclusions are also occurred in chalcopyrite.

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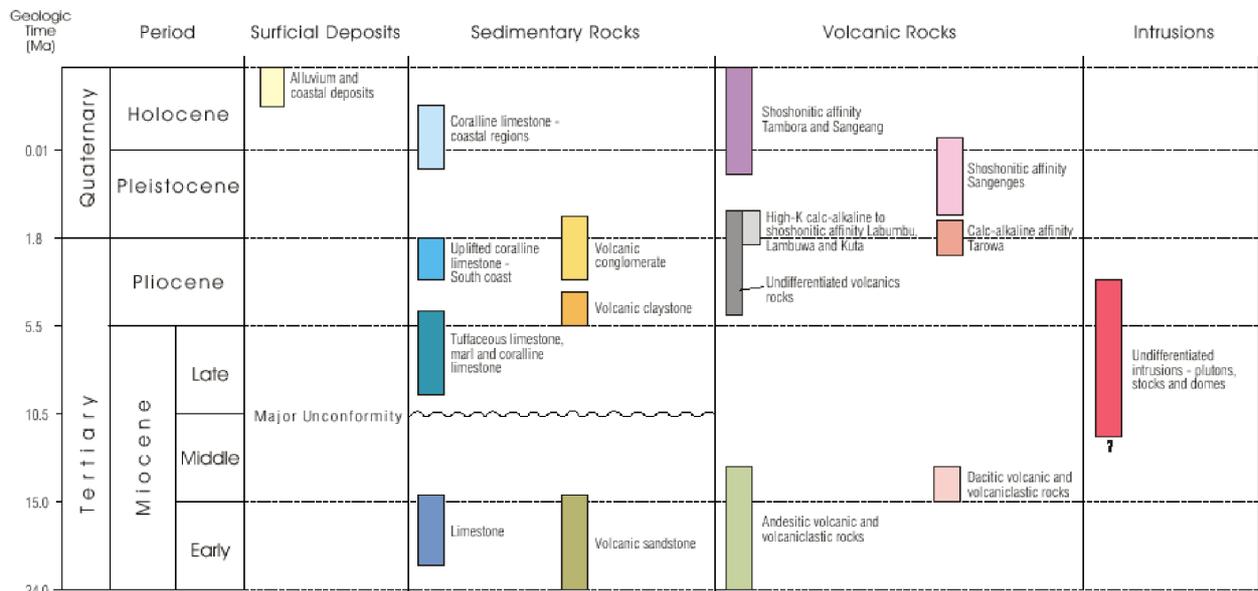
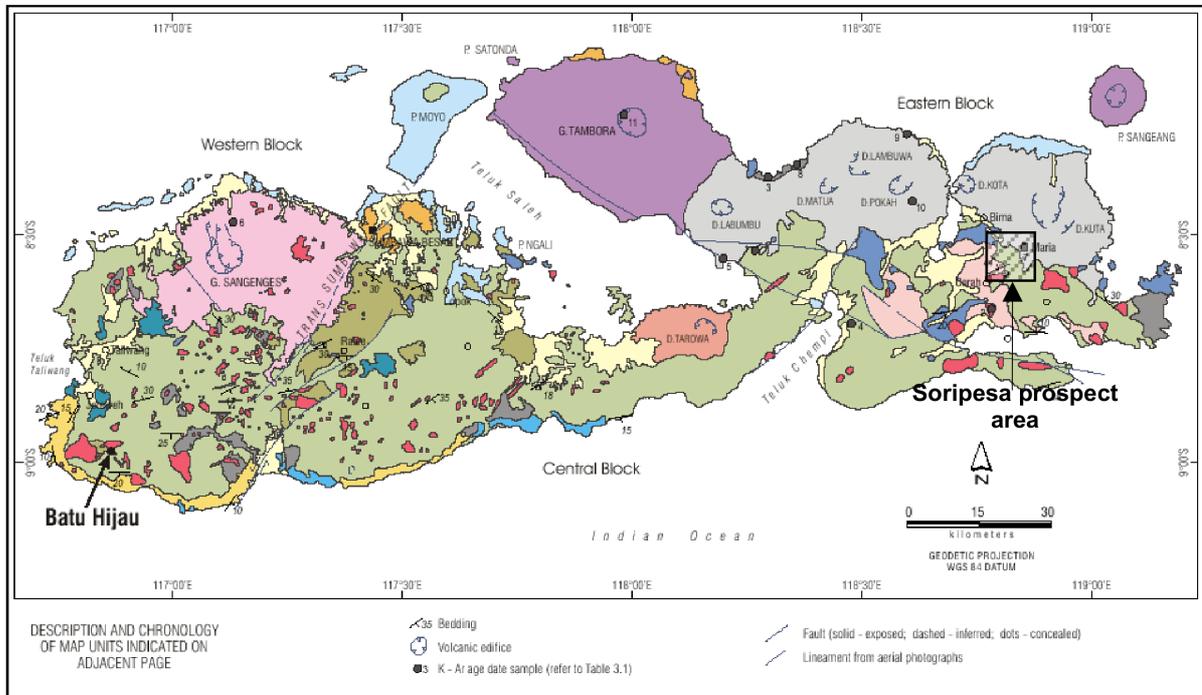
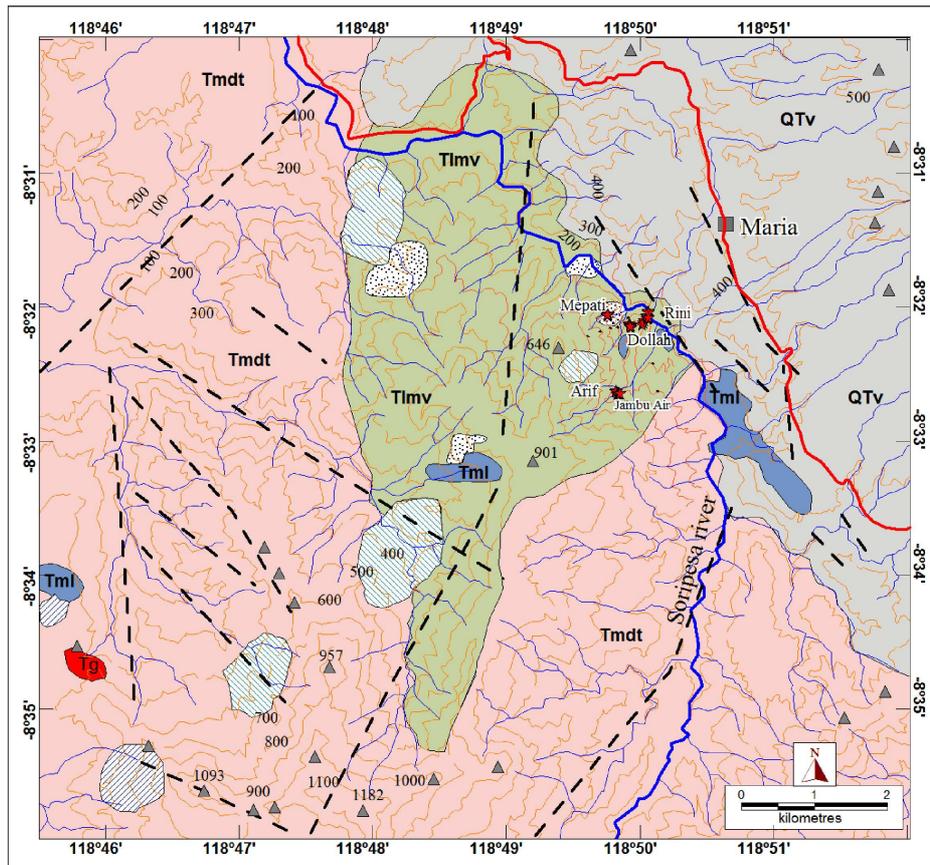


Figure 1: Simplified geologic map and chronology of map units for Sumbawa island, Indonesia including Soripesa prospect area (Source: Garwin, 2002)



LEGEND

QTV	Quaternary agglomerate to breccias andesitic volcanic, not consolidated matrix supported with fragments
Tg	Andesite intrusive
Tmdt	Dacitic volcaniclastic, agglomeratic to breccias gradation to fine-grained tuff
Tml	Fossiliferous limestone
Tlmv	Andesitic volcaniclastic, agglomeratic to breccias gradation to fine-grained tuff
	Highly silicified + pyrite±manganese alteration in places silica has been leached out (gossanous) and contains highly oxidized
	Clay-quartz alteration
	Chlorite-epidote alteration
★	Quartz vein with base metals mineralization (azurite, chalcopyrite, malachite, galena, sphalerite, pyrite)
- - -	Fault
~	River and stream
—	Motor road
—	Contour (m)
▲	Spot high

Figure 2: Geological map of around Soripesa prospect area with epithermal quartz veins (modified after Noya *et al*, 2009)

Pyrite. most of pyrite occurs as euhedral form and some are replaced by chalcopyrite. Pyrite occurs in both host rock samples and vein samples. Some are replaced by Fe oxide minerals such as goethite and hematite which are show in concentric texture (Figure 3a). Some pyrite minerals show ribbon texture.

Galena. galena is one of common ore minerals in this research. Some of galena shows a typical texture of triangular pits (Figure 3c). Most of the crystal outlines of galena are generally replaced by chalcocite (Figures 3c and 3e).

Sphalerite. sphalerite is also a common ore mineral of polymetallic epithermal quartz veins in this research area. Most of sphalerite show intergrowth texture with oriented an unoriented inclusions of chalcopyrite (the so-called "chalcopyrite disease"; Barton, 1978). Traditionally, the texture has been described as an exsolution of chalcopyrite from sphalerite. However Barton (1970, 1978) and Kalogeropoulos (1982) have convincing evidence that "chalcopyrite disease" is a cancerous replacement produced by Cu in aqueous solution reacting with FeS in sphalerite. In some case, the outlines of sphalerite minerals are in general concave toward quartz minerals (Figure 3d). It means that sphalerite is younger than quartz in order of deposition. Sphalerite shows a lot of cracks which are replaced by chalcocite and covellite

Chalcocite. they are found as replacement minerals and replace sphalerite, chalcopyrite, and galena (Figures 3b-e). In some case, they show colloform textures.

Covellite. they also occur as replacement minerals and replaced mainly in chalcocite. Azurite and malachite are found as hydrothermal alteration minerals. Some azurite shows zonal texture under ore microscopy (Figure 3e).

4.2 Ore-gangue Minerals Paragenesis

The paragenesis for a mineral deposit or a genetically related group of deposits is inferred

from the mutual relationships among the minerals and assemblages (Misra, 1999). The interpretation of microscopic textures of ore minerals is very useful to prepare the paragenesis diagrams of those ore minerals. The paragenesis was made from a careful study of polished sections and thin sections. Textures of ore minerals such as banded, exsolution, replacement, and zone, have been interpreted to correspond to the order of deposition. Pyrite and quartz minerals show their euhedral crystal form. It is clear that sphalerite is younger than quartz in order of deposition (Figure 3d). In Figure 3e, it can be seen clear that covellite replace chalcocite, and chalcocite commonly coats and replaces the margins of galena and sphalerite, and sphalerite replace galena. Chalcopyrite inclusions also occur as exsolution textures in crystallographic direction of sphalerite (Figure 3e). Fe oxide minerals replace in the cracks of chalcopyrite and pyrite (Figures 3a and 3f). Azurite and malachite occurs as later phase minerals.

Based on the textural characteristics, order of deposition can be divided in to two phases as early phases and later phases. Quartz, galena, pyrite, sphalerite, and chalcopyrite are formed in early stage of deposition. Chalcocite, covellite, azurite, malachite, and Fe oxide minerals are formed later phase of ore deposition. The paragenesis of a mineral deposit can provide a record of the local changes in geochemical parameters (T, P, fluid composition, etc.) in the course of mineralization (Bastin, 1955). The paragenesis diagrams of ore and gangue minerals from polymetallic epithermal quartz veins at Soripesa prospect area are shown in Figure 4.

4.3 Mineral Chemistry of Pyrite, Galena, and Sphalerite

Minor Elements in Pyrite

Pyrite is one of the most abundant minerals in the polymetallic epithermal at Soripesa prospect area. 21 pyrite minerals are detected by SEM-EDX to know the elemental composition. Average contents (wt.%) of some minor elements of pyrite are Cd (0.2), Co (0.45), Ni (0.14), Ag (0.24), Au (1.42), and As (0.75) (Table 1). Some pyrite minerals under SEM-EDX show

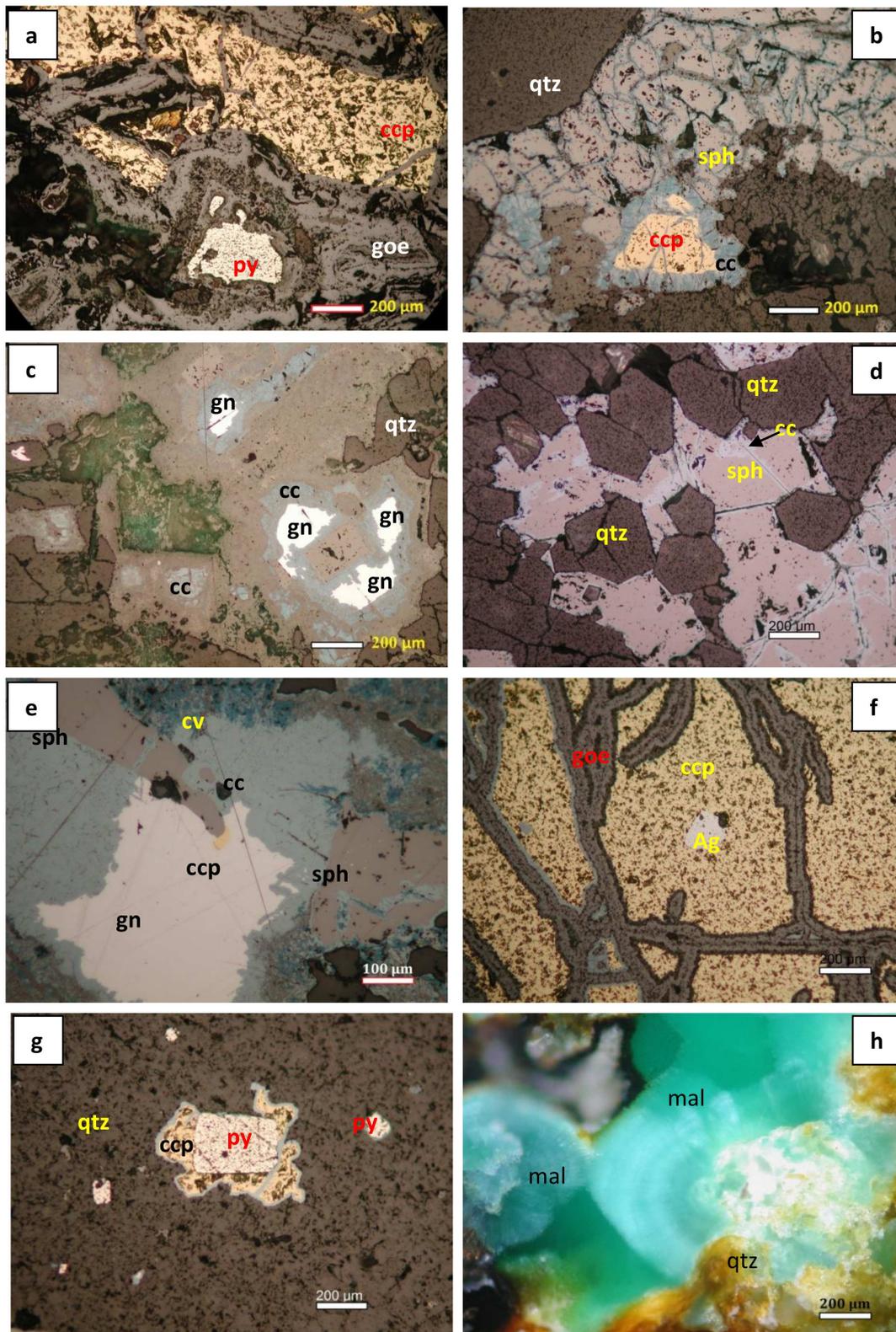


Figure 3: Photomicrographs showing ore mineral assemblages of polymetallic epithermal quartz veins at Soripesa prospect area (qtz = quartz, ccp = chalcopyrite, gn = galena, Ag = silver, py = pyrite, cc = chalcocite, mal = malachite, and goe = goethite).

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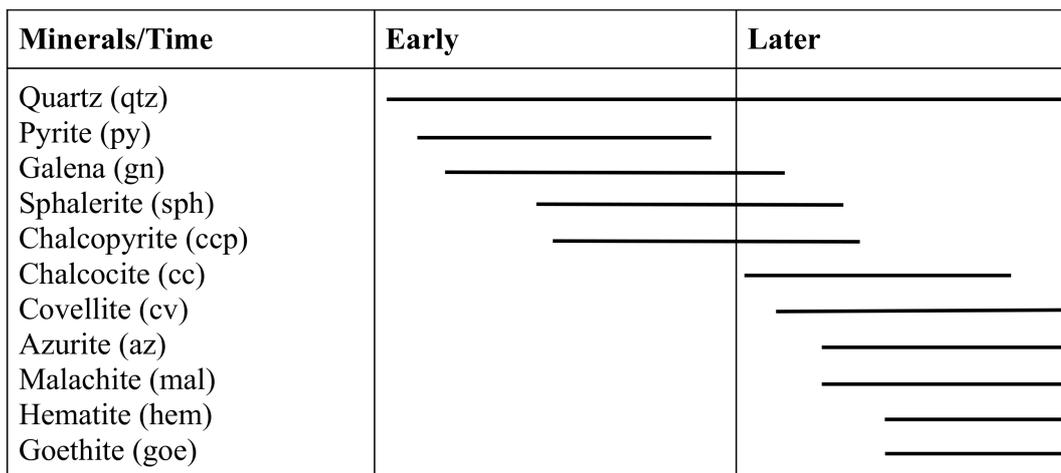


Figure 4: Paragenetic diagram of ore-gangue minerals assemblage of polymetallic epithermal quartz veins at Soripesa prospect area.

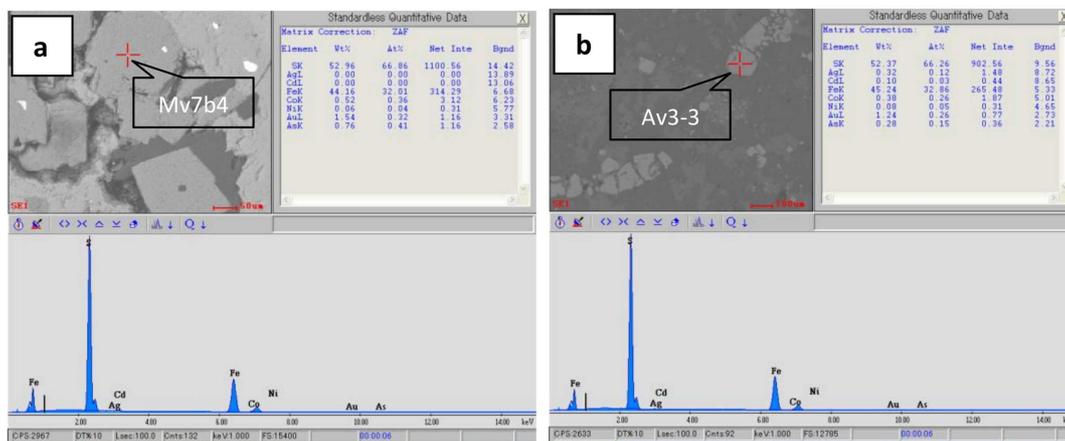


Figure 5: Elemental composition and peak intensity of some pyrite minerals detected by SEM-EDX analyses.

Table 1: SEM-EDX analyses of element composition (wt.%) of Pyrite minerals. Element composition (normalized wt.%) of Pyrite minerals.

No.	Samples	Fe	S	Cd	Co	Ni	As	Ag	Au	Co/Ni
1	Av3-1	43.57	52.25	0.26	0.51	0.20	1.05	0.25	1.74	2.55
2	Av3-2	44.19	53.41	0.24	0.20	0.09	0.61	0.24	1.01	2.22
3	Av3-3	45.24	52.37	0.10	0.38	0.08	0.28	0.32	1.24	4.75
4	Dv10-1	43.08	52.11	0.38	0.72	0.35	1.03	0.28	2.05	2.06
5	Dv10-2	45.92	51.17	0.28	0.53	0.14	0.95	0.13	0.87	3.79
6	Dv10-3	44.57	52.98	0.22	0.36	0.17	0.62	0.33	0.75	2.12
7	Dv10-5	44.35	52.51	0.18	0.52	0.19	0.56	0.13	1.56	2.74
8	JA5-2	44.50	52.02	0.33	0.39	0.14	0.87	0.33	1.42	2.79
9	Mv7a1	43.71	51.18	0.31	0.65	0.25	0.90	0.49	2.51	2.60
10	Mv7a2	45.11	52.29	0.15	0.44	0.19	0.62	0.20	1.00	2.32
11	Mv7a3	44.56	52.66	<0.01	0.36	0.21	0.78	<0.01	1.42	1.71
12	Mv7a4	47.83	48.78	<0.01	0.29	0.13	0.86	<0.01	2.11	2.23
	Avg	44.60	52.19	0.20	0.45	0.14	0.75	0.24	1.42	2.81

Table 2: Geometric means of cobalt (Co) and nickel (Ni) contents (in ppm) and Co:Ni ratios of sediment, hydrothermal, and massive sulphide pyrite minerals (Xuexin, 1984).

Type of pyrite	Co	Ni	Co:Ni
This paper	4500	1400	2.81
Sedimentary	41	65	0.8
Volcano-hydrothermal, metamorphosed and skarn-hydrothermal	141	121	2–3
Volcanogenic Massive Sulphide	486	56	3.5

their image, result of element composition (normalized wt.%) and peaks of elements (Figure 4).

Xuexin (1984) mentioned a statistically significant difference among sedimentary, hydrothermal (replacement veins), and massive sulfide pyrites (Table 2). According to his study, the massive sulfide pyrites are characterized by a Co:Ni ratio between 5 and 50; hydrothermal pyrites by a variable Co:Ni ratio (often less than 5); and sedimentary pyrites by a much lower Co:Ni ratio (typically <1). Pyrite from sulfide ore deposits generally contains appreciable amounts of cobalt and has $Co > Ni$.

According to Carstens (1942), pyrite of sedimentary origin is characterized by containing less than 100 ppm cobalt and $Co < Ni$, whereas pyrite of hydrothermal origin has 400-2400 ppm cobalt and $Co > Ni$. Pyrite from high-temperature deposits is generally high in cobalt, but noted exceptions and considered that the effect of temperature was slight. In this research, average content of Co (0.45 wt.%) is higher than Ni content (0.14 wt.%) and it means that their origin may be hydrothermal origin. Pyrite of high-temperature and hydrothermal sulfide deposits has only a slightly higher average nickel content than pyrite of sedimentary origin and the ranges of concentration overlap. Hawley (1952) found that Ni content was a little higher in high-temperature samples. Average ratio for Co/Ni is 2.81. The Co/Ni ratios for volcano-hydrothermal, metamorphosed volcanogenic and skarn-hydrothermal pyrites are

in the range of 2–3 (Xuexin, 1984). It means that pyrites from Soripesa prospect area may be formed under the hydrothermal condition.

There is a correlation between As and S, and no correlation between As and Fe in pyrite of this research (Figures 7a-b). These relations indicate substitution of As for S in the pyrite structure, as would be expected for pyrite containing As^{1-} (Deditius *et al*, 2009). There is also a correlation between Au and Ag but no correlation between Au and As (Figures 6a-b).

Minor Elements in Galena

Galena is also common ore minerals of epithermal quartz veins in this research area. 16 galena minerals were detected by SEM-EDX for their mineral chemistry. Some detected minor elements of galena from this research are Fe (1.4 wt.%), Hg (1.92 wt.%), Ag (0.07 wt.%), Au (1.48 wt.%), Sb (0.39 wt.%), and Se (1.05 wt.%) (Table 3). SEM image, peaks of elements, and results of element composition (normalized wt.%) of some Galena analysed by SEM-EDX are shown in Figure 8.

Tischendorf (1955) found that the antimony content (present at least in part as tetrahedrite) decreased with decreasing temperature of formation. Schroll (1955) considers the presence of bismuth in galena to be a mark of high-temperature magma-near deposits. Galena in this research has below detection limit of bismuth. Schroll (1955) found that the content of silver varied directly with those of antimony and bismuth and inversely with content of arsenic. Fleischer (1956) reported that the content of silver, bismuth, and antimony are decreased with decreasing temperature of formation. Silver content (0.07) is very low and it means that galena from this research may be formed in decreasing temperature of formation.

Minor Elements in Sphalerite

Sphalerite is one of the most useful indicators of the environment of ore deposition because of its refractory nature, wide distribution in natural environments, and wide range of composition resulting from substitution of Zn by Fe (Misra, 1999). Detected minor elements in spha-

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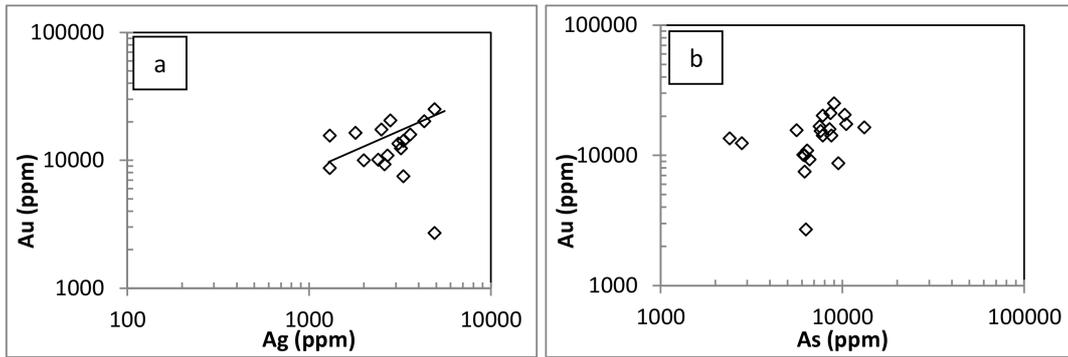


Figure 6: Plots showing correlation diagrams of Au, Ag, and As concentration in pyrite.

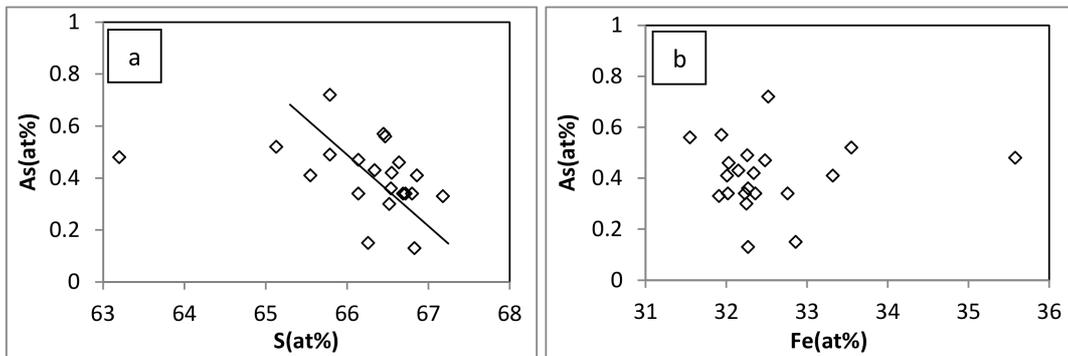


Figure 7: Plots showing correlation diagrams of As, S, and Fe concentration in pyrite.

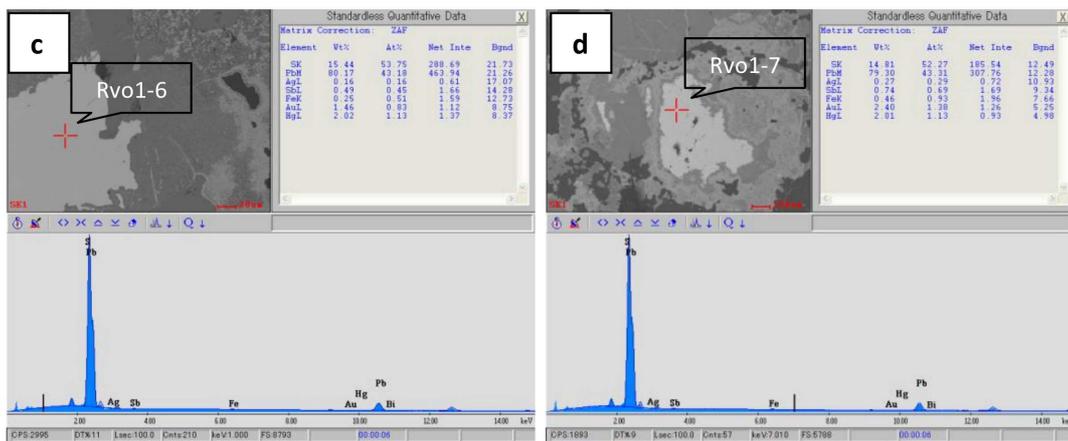


Figure 8: Elemental composition and peak intensity of some galenas detected by SEM-EDX analyses.

Table 3: SEM-EDX analyses of element composition (wt.%) of galena. Element composition (normalized wt.%) of Galena minerals.

No.	Samples	Pb	S	Fe	Hg	Sb	Ag	Au	Se
1	Av3-4	77.70	14.16	3.97	1.24	0.42	<0.01	1.39	0.95
2	Av3-5	76.03	14.00	3.84	2.47	0.54	0.23	1.15	1.55
3	Av3-6	77.05	14.33	3.12	2.30	0.13	<0.01	2.04	1.03
4	Av3-7	78.17	14.58	3.49	1.51	0.32	<0.01	1.27	0.66
5	Av3-8	77.86	14.68	3.82	1.11	0.51	<0.01	0.97	0.36
6	Av3-15	77.89	14.52	0.33	2.64	0.48	<0.01	1.88	1.35
7	Av3-16	78.32	14.72	0.34	2.26	0.28	<0.01	1.46	1.72
8	Av3-17	80.31	15.52	0.33	0.91	0.18	<0.01	0.56	0.81
9	Rvo1-6	80.17	15.44	0.25	2.02	0.49	0.16	1.46	n/a
10	Mv4-1	79.56	15.17	0.37	2.66	0.22	<0.01	2.02	n/a
11	Mv4-2	79.85	14.93	0.34	2.46	0.71	<0.01	1.71	n/a
12	Rvo1-7	79.30	14.81	0.46	2.01	0.74	0.27	2.40	n/a
	Avg	79.02	14.83	1.41	1.92	0.39	0.07	1.48	1.05

(n/a = not analysed)

lerite are Fe (1.04 wt.%), Ga (0.7 wt.%), Ge (0.54 wt.%), Cd (0.7 wt.%), and Ag (0.2 wt.%). Au content is below detection limit in all sphalerite (Table 4). Figure 9 show the result of some sphalerite minerals analysed by SEM-EDX. The results show SEM image, peaks of elements, and elemental composition in wt.%.

Jonasson and Sangster (1978) investigated the Zn/Cd ratios of sphalerites from some sulphide ores in Canada and concluded that the Cd contents and Zn/Cd ratios in sphalerites vary with the genetic types of deposit. The Zn/Cd ratios (average 93.34) in sphalerite from the Soripesa prospect area are close to Zn/Cd ratios (104–214) of sphalerites from hydrothermal deposits and skarn-hydrothermal deposits. Stoiber (1940) and Schroll, (1955) considered that the gallium content is most likely to be high in sphalerites from low-temperature deposits such as those of Mississippi Valley type and in those from low-temperature quartz veins.

In this research, average Ga content (0.7 %) in sphalerites is very high and it means that they may be formed under low temperature condition. Many investigators agree that sphalerite from low-temperature deposits to be higher in germanium content than those from mesothermal or high-temperature deposits (Stoiber, 1940; Warren and Thompson, 1945, Fleischer, 1999). In this research, average content of Ge (0.54%) is also very high. Vaughan and Craig (1997) suggested that a zinc concen-

trate from a pyritic ore may contain 61-63 wt.% Zn, but a concentrate from a pyrite-pyrhotite ore will often contain only 51-58 wt.% zinc and be considerably less valuable. The average content of Zn (63.48 wt.%) in sphalerite of this research may be related with pyritic ore condition.

Sphalerite Geothermometry and Geobarometry

A very interesting recent development is the Ga/Ge geothermometer using sphalerite (Moller, 1985). Ga/Ge can be used to determine temperatures in the source regions of ore solutions and to estimate the degree of mixing of hot parental ore fluids with cool, near surface waters (Evans, 1993). Ten samples were simultaneously detected levels for these two elements. The application of these geothermometer results of calculating the logarithm follows:

$$\log[(\text{Ga/Ge})_f \log[(\text{Ga/Ge})_{\text{sph}}]] \quad (1)$$

Note that: f = mineraliser fluid; sph = sphalerite.

Determining the values of log (Ga/Ge) for the analyzes of sphalerite minerals, we obtain the values in the ranges between -0.4 to 1.15. These values are applied to the chart geothermometer Ga/Ge, based on geothermometer of Al/Si, and in this case that of muscovite-chlorite-quartz, given the degree of metamor-

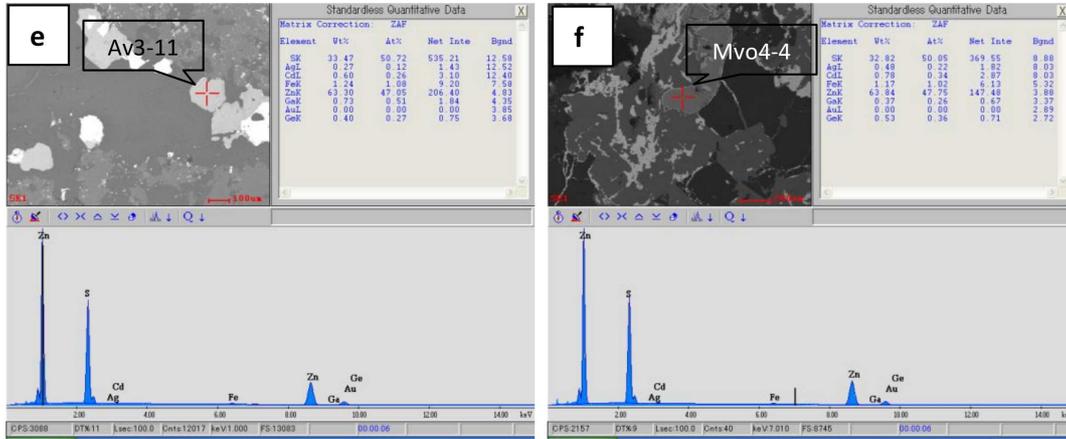


Figure 9: Elemental composition and peak intensity of some sphalerites detected by SEM-EDX analyses.

Table 4: Chemical SEM-EDX analyses of element composition (wt.%) of sphalerites.

No.	Samples	Zn	S	Fe	Ga	Ge	Cd	Ag	Au	Zn/Cd
1	Av3-9	62.82	34.14	1.08	0.68	0.29	0.65	0.34	<0.01	96.65
2	Av3-10	63.22	33.81	1.21	0.71	0.54	0.50	<0.01	<0.01	126.44
3	Av3-11	63.30	33.47	1.24	0.73	0.40	0.60	0.27	<0.01	105.50
4	Av3-12	62.29	35.27	0.56	0.65	0.36	0.63	0.25	<0.01	98.87
5	Rvo1-1	64.31	32.62	1.11	0.50	0.48	0.64	0.34	<0.01	100.48
6	Rvo1-2	63.81	33.21	0.93	0.83	0.25	0.71	0.26	<0.01	89.87
7	Rvo1-3	63.38	33.38	0.84	0.59	0.71	0.83	0.27	<0.01	76.36
8	Rvo1-4	64.07	32.00	1.10	0.92	0.65	0.86	0.40	<0.01	74.50
9	Rvo1-5	63.79	32.24	1.17	1.06	0.84	0.77	0.12	<0.01	82.84
10	Mvo4-4	63.84	32.84	1.17	0.37	0.92	0.78	0.48	<0.01	81.85
	Avg	63.48	33.30	1.04	0.70	0.54	0.70	0.27	<0.01	93.34

phism. The chart shows the values of temperatures between 180 to 240°C. Figure 10 presents the projection of the respective analysis chart that correlates the ratio Ga/Ge with the temperature. This graph is based on geothermometers known systems of Al/Si and published data for reasons of Ga/Ge in many geological systems (Möller, 1985; 1987).

Neiva (1956) considers that the presence of Ge is higher in sphalerites that are formed at low temperature. Based on the fluid inclusion data from quartz (host minerals), ranges of homogenization temperature for quartz veins is between 180 and 300°C (Khant *et al*, 2012a). The temperature differences observed between the two geothermometers can be related to the mixing and circulation of fluids late and/or pre-fluid surface (Möller, 1985; 1987).

Kullerud (1953) suggested that the FeS content of sphalerite gave a direct measurement of its temperature of deposition. Average Fe content (1.04 %) of sphalerite in this research is very low. Mole % FeS in sphalerite is also very low and between 1–2 %. In Zn-Fe-S of low pressure system, this value (1–2 %) falls in pyrite field (Figure 11). Within the pyrite field the decrease in FeS content of sphalerite with increasing a_{S_2} (sulfur activity) is much greater resulting in a close spacing of isopleths near the pyrite-pyrrhotite buffer and in very low FeS contents in sphalerite at high, a_{S_2} (10^{-10} to 10^{-11}). It is commonly found that sphalerites formed at high a_{S_2} have a honey yellow to light brown colour with their low FeS contents whereas sphalerites formed at low a_{S_2} and within the pyrrhotite field are dark brown to black (Scott, 1983). Those factors mean they may be formed in low temperature condition.

5 Conclusion

Common ore minerals of the polymetallic epithermal quartz veins at Soripesa prospect area are chalcopyrite ($CuFeS_2$), galena (PbS), sphalerite (ZnS), azurite, malachite, chalcocite, covellite, pyrite (FeS_2), hematite, and other Fe oxide minerals. Common textures of ore minerals are replacement, exsolution, banded (colloform-crustiform), and zone textures. The paragenesis

diagram was drawn from a study of polished sections and thin sections under microscope. Ore mineral textures have been interpreted to identify the order of deposition and to make paragenesis diagram. Quartz, pyrite, galena, sphalerite and chalcopyrite are formed in early stage and chalcocite, covellite, azurite, malachite, and Fe oxide minerals are formed in late stage of order of deposition.

Pyrite, galena, and sphalerite are analysed by SEM-EDX to know the elemental compositions. Average contents (wt.%) of some minor elements of pyrite are Cd (0.2), Co (0.45), Ni (0.14), Ag (0.24), Au (1.42), and As (0.75). Some detected minor elements of galena from this research are Fe (1.4 wt.%), Hg (1.92 wt.%), Ag (0.07 wt.%), Au (1.48 wt.%), Sb (0.39 wt.%), and Se (1.05 wt.%). Detected minor elements in sphalerite are Fe (1.04 wt.%), Ga (0.7 wt.%), Ge (0.54 wt.%), Cd (0.7 wt.%), and Ag (0.2 wt.%). Average content of Co (0.45) is higher than Ni content (0.14) and average ratio for Co/Ni is 2.81 in pyrite from this research. It means that origin of pyrite from this research may be hydrothermal origin. Based on the low contents of Sb, Bi, and Ag in average, we can conclude that galena from this research are formed in decreasing temperature of formation.

The Zn/Cd ratios (average 93.34) in sphalerite from the Soripesa prospect area are close to Zn/Cd ratios (104–214) of sphalerites from hydrothermal deposits. Low content of Fe, and high content of Ga and Ge, also indicate that sphalerites from this research are formed in low temperature quartz veins. Based on Ga/Ge geothermometer, the formation temperatures of sphalerites can be estimated and they may be formed at between 180 and 240°C. 1 to 2 Mole % FeS in sphalerite indicate that ore deposition formed under high a_{S_2} (10^{-10} to 10^{-11}) and associated with pyrite field.

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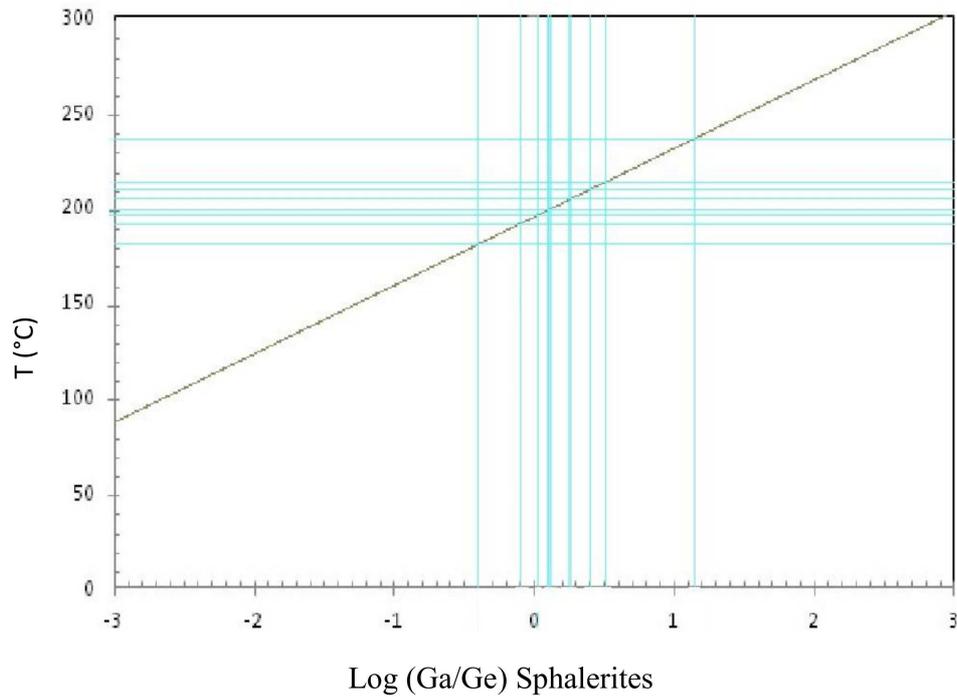


Figure 10: Graph showing the dependence between Ga/Ge ratios in sphalerite and formation temperature. This graph is based on existing Al/Si geothermometers. The musc-clo-qtz line is the chlorite geothermometer in the Si/Al system (Möller, 1985; 1987).

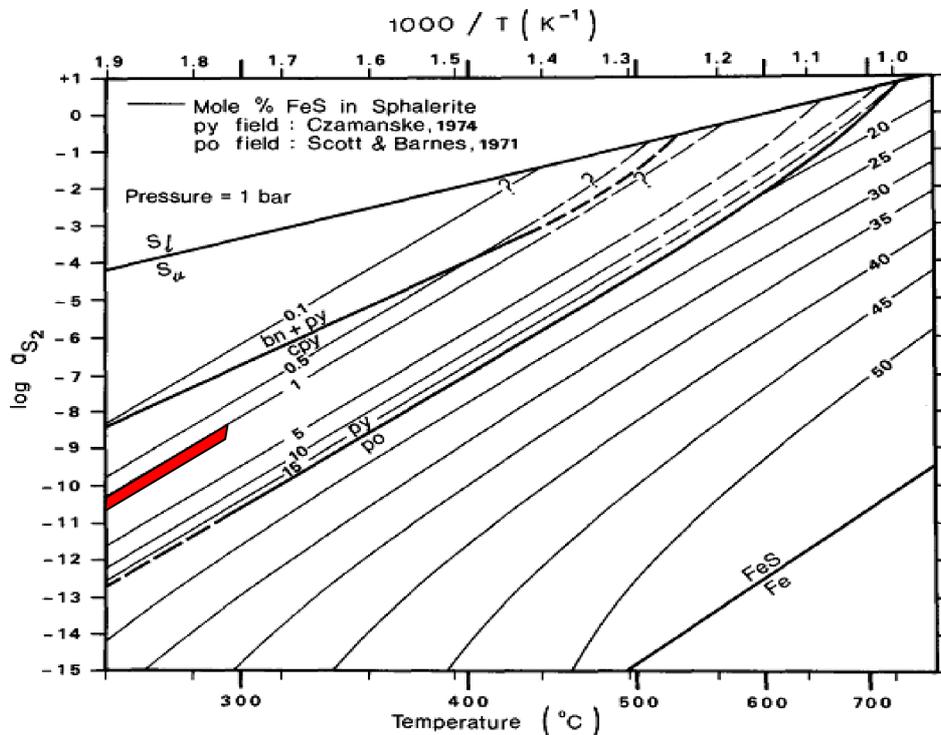


Figure 11: Phase relationships for the Fe-Zn-S system at 1 bar compiled from Barton and Toulmin (1966), Scott and Barnes (1971) and Czamanske (1974). Red line represents the sulfur activity of sphalerite in this research. Abbreviations: bn = bornite; cpy = chalcopyrite; po = pyrrhotite; py = pyrite; S_l = liquid sulphur; S_v = sulphur vapour.

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